$\qquad$ Date $\qquad$

## Density Problems

1. Write the equation for Density: $\qquad$
Unit for Solids $\qquad$
Unit for Liquids $\qquad$
2. Place the symbols, from the density equation, in their proper place in the triangle:

3. Write the equation for Volume: $\qquad$
Unit for Solids $\qquad$
Unit for Liquids $\qquad$
4. Write the equation for Mass: $\qquad$ Unit $\qquad$

## Example:

1. Iron has a density of $7.86 \mathrm{~g} / \mathrm{cm}^{3}$. What would the mass of an iron bar with a volume of $40 \mathrm{~cm}^{3}$ ?
Density $\quad 7.86 \mathrm{~g} / \mathrm{cm}^{3}$

Volume $\qquad$ Mass $\qquad$
Formula With Knowns Answer with unit
$\mathrm{M}=\mathrm{DV}=(7.96)(40)=0.199 \mathrm{~g}$
(Show all work and include all units - Just like in the example problem)

1. A solid ball has a mass of 50 grams and a volume of $20 \mathrm{~cm}^{3}$. What is the density?
$\qquad$ Volume $\qquad$ Mass $\qquad$
Formula
With Knowns
Answer with unit
2. Silver has a density of 10.5 grams $/ \mathrm{cm}^{3}$. What would be the volume of a silver coin if it had a mass of 5 g ?

Density $\qquad$
Formula
Volume $\qquad$ Mass $\qquad$
Answer with unit
3. Given that the density of gold is $19.3 \mathrm{~g} / \mathrm{cm}^{3}$, what would be the mass of a gold bar that had a volume of $100 \mathrm{~cm}^{3}$ ?

Density $\qquad$ Volume $\qquad$
With Knowns
Answer with unit
4. A rock has a volume of $30 \mathrm{~cm}^{3}$ and a mass of 60 g . What is its density?
Density $\qquad$

Volume $\qquad$ Mass $\qquad$
Formula
With Knowns
Answer with unit
5. Bronze has a density of $9.87 \mathrm{~g} / \mathrm{cm}^{3}$. What would be the volume of a bronze coin if it had a mass of 5 g ?
$\qquad$ Volume $\qquad$ Mass $\qquad$
Formula
With Knowns
Answer with unit
6. Freshwater has a density of $1.0 \mathrm{~g} / \mathrm{cm}^{3}$. What volume container would be needed to hold 50 g of this water?
Density $\qquad$
Formula

Volume $\qquad$ Mass $\qquad$
With Knowns
Answer with unit
7. Aluminum has a density of $2.7 \mathrm{~g} / \mathrm{cm}^{3}$. What would be the mass of a bar of aluminum that had a volume of $100 \mathrm{~cm}^{3}$ ?
$\qquad$
Density
Volume $\qquad$ Mass $\qquad$
With Knowns
Answer with unit
8. A block of metal has a volume of $15 \mathrm{~cm}^{3}$ and a mass of 45 g . What is its density?
$\qquad$
Density
Volume $\qquad$ Mass $\qquad$
Formula
With Knowns
Answer with unit
9. The water in the Great Salt Lake in Utah can reach a density of $1.25 \mathrm{~g} / \mathrm{cm}^{3}$. What volume container would be needed to hold 50 g of this water?

Density $\qquad$ Volume $\qquad$
Formula

With Knowns
Answer with unit
10. What is the density of Carbon dioxide gas if 0.196 g occupies a volume of 100 ml ?

Density $\qquad$ Volume $\qquad$
Formula
With Knowns With

Mass $\qquad$
Answer with unit

