Name \_\_\_\_\_ Date \_\_\_\_\_

# **Properties of Water Lab**

**Goal:** To investigate the various properties of water.

# **Vocabulary:**

HydrophilicAdhesion

Hydrophobic

Capillary Action

Cohesion

**Task Two** 

Surface Tension

## **Task One: Paper Towel in Water**

- 1. Fill a clear plastic cup with 25 mL of water.
- 2. Place 4 drops of food coloring into the cup and stir with a small wooden stick.
- 3. Tape a strip of paper towel to a pencil.
- 4. Balance the pencil on top of the cup of colored water and let the paper towel strip hang down and touch the top of water.
- 5. Observe and record happens to the paper towel after a period of 3 minutes.
- 6. Throw away the paper towel and rinse out the plastic cup.
  - What did you observe?
  - Which property or properties of water were displayed? Circle all that apply:

Cohesion Adhesion Capillary Action Hydrophobic Hydrophilic

Universal Solvent Surface Tension

Pencil Paper towel

Beaker filled with water

#### Task Two: Salt and Water Mixture

- 1. Place 150 mL of water in a clear plastic cup.
- 2. Add a teaspoon of salt to the water and stir.
- 3. Observe and record what happens after you have stirred the water solution for 1 minute.
- 4. Rinse out the cup when you are finished.
  - What did you observe?
  - Which property or properties of water were displayed? Circle all that apply:

**Task One** 

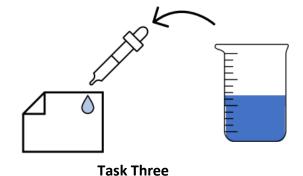
Cohesion Adhesion Capillary Action Hydrophobic Hydrophilic

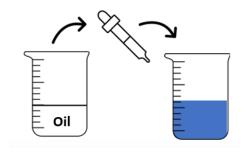
Universal Solvent Surface Tension

#### **Task Three: Water Droplet on Wax Paper**

- 1. Use an eyedropper to place a small drop of water onto a piece of wax paper.
- 2. Carefully tilt the wax paper back and forth.
- 3. Observe and record what happens to the water droplet.
- 4. Throw the piece of wax paper away, when you are finished.
  - What did you observe?
  - Which property or properties of water were displayed? Circle all that apply:

Cohesion Adhesion Capillary Action Hydrophobic Hydrophilic
Universal Solvent Surface Tension





Task Four

#### Task Four: Water and Oil

- 1. Fill a clear plastic cup with 200 mL of water.
- 2. Add 3 drops of food coloring into the water and use a stirring stick to fully dissolve the food coloring.
- 3. Use a plastic pipette to add 10 drops of vegetable oil to the colored water.
- 4. Stir the mixture with the wooden stick, then let the mixtures sit for 1 minute.
- 5. Observe and record what happens.
- 6. Throw away the paper cup, wooden stick, and plastic pipette.
  - What did you observe?
  - Which property or properties of water were displayed? Circle all that apply:

Cohesion Adhesion Capillary Action Hydrophobic Hydrophilic

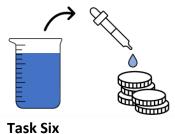
Universal Solvent Surface Tension

## Task Five: Paperclip on Water

- 1. Fill a cup with water.
- 2. Carefully place a small paperclip on top of the water so that it floats on the water.
- 3. If you are unsuccessful, try again several times, altering your technique.
- 4. If you are still unsuccessful, after several attempts, reshape the paperclip and try again.
- 5. Carefully place the larger paperclip on top of the water so that it floats on the water.
- 6. Record your observations:
  - Did you need to reshape your paperclip to get it to float?
  - Why do you think the new shape was easier or more difficult to float?
  - Which was easier to float, the smaller or larger paperclip? Why do you think this was?
  - Which property or properties of water were displayed? Circle all that apply:

Cohesion Adhesion Capillary Action Hydrophobic Hydrophilic
Universal Solvent Surface Tension





#### Task Six: Drops of Water on a Penny

- 1. Guess how many drops of water and fit onto a penny?
- 2. Use the eyedropper to place drops of water onto a penny, counting each drop of water as you go.
- 3. Stop when the water flows over the edge of the penny.
- 4. Record the number of drops of water you successfully placed onto the penny.
  - Number of drops of water placed on the penny:
  - Which property or properties of water were displayed? Circle all that apply:

Cohesion Adhesion Capillary Action Hydrophobic Hydrophilic

Universal Solvent Surface Tension