

Name _____ Date _____

Isotope Review - Calcium

1. How many protons are in Calcium - 40? _____
2. How many electrons are in Calcium – 40 ? _____
3. How many neutrons are in Calcium – 40 ? _____
4. Draw the isotope symbol for Calcium-40

5. How many protons are in Calcium – 42? _____
6. How many neutrons are in Calcium – 42? _____
7. Draw the isotope symbol for Calcium-42:

8. Which is the most abundant Calcium isotope?

Name _____ Date _____

Isotope Review - Sodium

1. How many protons are in Sodium - 22? _____
2. How many electrons are in Sodium – 22 ? _____
3. How many neutrons are in Sodium – 22 ? _____
4. Draw the isotope symbol for Sodium- 22

5. How many protons are in Sodium – 23? _____
6. How many neutrons are in Sodium – 23? _____
7. Draw the isotope symbol for Sodium- 23:

8. Which is the most abundant Sodium isotope?

Name _____ Date _____

Isotope Review - Bromine

1. How many protons are in Bromine - 80? _____
2. How many electrons are in Bromine – 80 ? _____
3. How many neutrons are in Bromine – 80 ? _____
4. Draw the isotope symbol for Bromine- 80

5. How many protons are in Bromine – 79? _____
6. How many neutrons are in Bromine – 79? _____
7. Draw the isotope symbol for Bromine- 79:

8. Which is the most abundant Bromine isotope?

Name _____ Date _____

Isotope Review - Chlorine

1. How many protons are in Chlorine - 35? _____
2. How many electrons are in Chlorine – 35 ? _____
3. How many neutrons are in Chlorine – 35 ? _____
4. Draw the isotope symbol for Chlorine- 35

5. How many protons are in Chlorine – 37? _____
6. How many neutrons are in Chlorine – 37? _____
7. Draw the isotope symbol for Chlorine - 37:

8. Which is the most abundant Chlorine isotope?