

Name _____

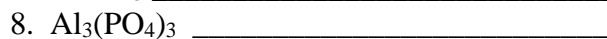
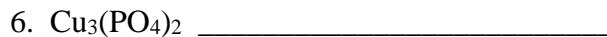
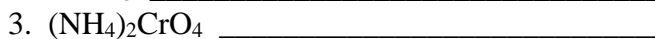
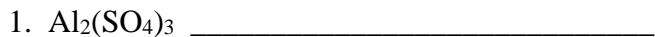
Date _____

Naming Binary Ionic Compounds

Ratio and Total Number of Atoms

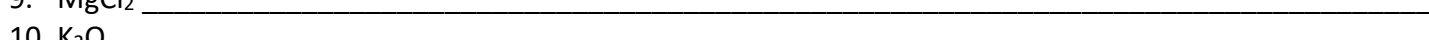
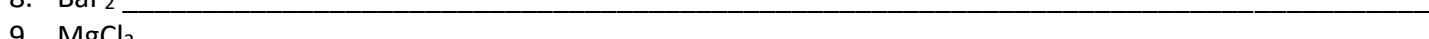
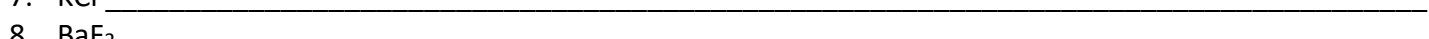
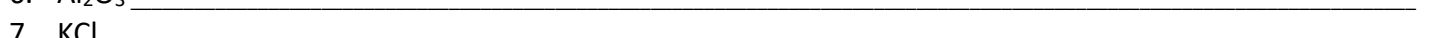
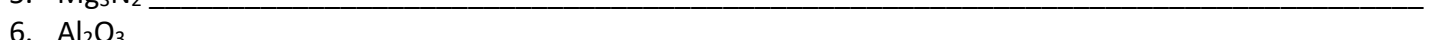
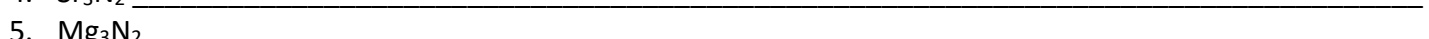
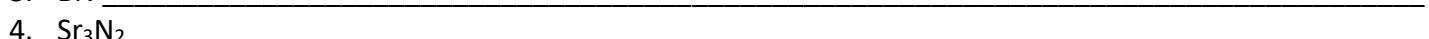
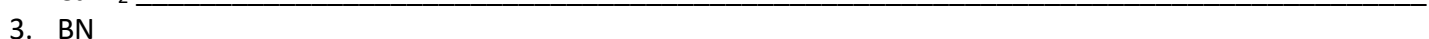
Give the ratio and the total number of atoms in the following compounds or molecules.

Example: $\text{Mg}(\text{NO}_3)_2 = 1 \text{ Mg} : 2\text{N} : 6\text{O} = 9 \text{ atoms}$



Naming Binary Ionic Compounds

Name the following binary ionic compounds.



Binary Ionic Compound Formulas

Write the formulas for the following binary ionic compounds

