

Name _____

Date _____

Polyatomic Ionic Compounds

Naming Compounds

1. $\text{Al}_2(\text{CO}_3)_3$ _____

2. $\text{B}_3(\text{PO}_4)_3$ _____

3. $(\text{NH}_4)(\text{NO}_3)$ _____

4. $(\text{NH}_4)_2\text{O}$ _____

5. $\text{Al}(\text{OH})_3$ _____

Writing Formulas

1. Ammonium Carbonate _____

2. Beryllium Chlorate _____

3. Ammonium Sulfate _____

4. Ammonium Oxide _____

5. Potassium Nitrate _____

6. Boron Phosphate _____

7. Ammonium Acetate _____

8. Calcium Chlorate _____

9. Aluminum Carbonate _____

10. Magnesium Acetate _____