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## pH Lab

## Directions

1. Write the name of each substance in the table below
2. Use the red litmus paper to determine whether each substance is a base or an acid and record your results in the table
3. Use the blue litmus paper to confirm your results and record your results in the table
4. Use the pH indicator paper to determine the exact pH of each substance and record your results in the table
5. Create a pH scale on the line, on back, and label weak acid, strong acid, weak base, strong base.
6. Write the name of each solution near its pH value on the line.

| Substance | Predicted pH | Red Litmus Paper | Blue Litmus <br> Paper | Actual pH | Acid or <br> Base |
| :--- | :--- | :--- | :--- | :--- | :--- |
| Distilled Water |  |  |  |  |  |
| Tap Water |  |  |  |  |  |
| Lemon Juice |  |  |  |  |  |
| Orange Juice |  |  |  |  |  |
| Tomato Juice |  |  |  |  |  |
| Dish Soap |  |  |  |  |  |
| Vinegar |  |  |  |  |  |
| Ammonia |  |  |  |  |  |
| Oven Cleaner |  |  |  |  |  |
| Coke |  |  |  |  |  |
| Tea |  |  |  |  |  |
| Baking Soda |  |  |  |  |  |

